Attorney Docket No.: 23085-08273 Client Ref: H10230237US01 (A02124)

USSN: 10/658,711

## IN THE UNITED STATES PATENT AND TRADEMARK OFFICE BEFORE THE BOARD OF PATENT APPEALS AND INTERFERENCES

APPLICANT:	Avetik Harutyunyan
APPLICATION NO.:	10/658,711
FILING DATE:	September 8, 2003
TITLE:	Methods For Preparation Of One-Dimensional Carbon Nanostructures
EXAMINER:	Kelly M. Stouffer
GROUP ART UNIT:	1792
ATTY. DKT. NO.:	23085-08273

## CERTIFICATE OF ELECTRONIC (EFS-WEB) TRANSMISSION

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Dated: September 29, 2008 By: /Narinder Banait/

Narinder S. Banait, Reg. No.: 43,482

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## **REPLY BRIEF**

This Reply Brief is filed in accordance with 37 CFR § 41.41 in response to the

Examiner's Answer, mailed on July 30, 2008.

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<u>Argument</u>

The Examiner's Answer mailed on July 30, 2008 is responsive to the Appeal Brief filed

on June 19, 2008. In the Examiner's Answer, the Examiner restates the rejections set forth in

the Final Office Action of August 21, 2007. The Appeal Brief filed on June 19, 2008 fully

addresses the deficiencies of these rejections.

In the Examiner's Answer, at page 8, the Examiner agrees with the applicant that

Muroyama teaches removing the oxide from the catalyst. The Examiner also appears to agree

with the applicant that Muroyama states that removal of the metal oxide provides for reliable

growth of the carbon film. However, the Examiner emphasizes that Muroyama only teaches

the removal of the <u>native oxide</u> for the more reliable growth of the carbon film, and does not

exclude the use of a purposely deposited metal oxide layer. If the presence of native oxide

prevents reliable growth of the carbon film, would one of skill in the art not expect a purposely

deposited metal oxide layer to also prevent reliable growth of the carbon film?

As the applicant had previously argued, Muroyama discloses three embodiments of

their methods for the synthesis of carbon films and the catalyst for use in all three embodiments

preferably removes the metal oxide on the surface of the organometallic compound thin layer.

Further, Muroyama discloses the use of nickel acetylacetonate as the metalorganic layer in

Examples 11 and 12, and in both examples, the corresponding metal oxide is removed. Thus,

the Muroyama embodiment that is closest to the applicants' claimed invention, the examples

show that the metal oxide is removed.

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The Examiner is using an incorrect test in maintaining the obviousness rejection. The

Examiner states:

Though Muroyama teaches that removing native oxide is preferred when using the metalorganic catalyst in paragraph 0095, it by no means requires

its removal, only prefers it for Muroyama's own device requirements.

The Examiner's Answer, page 8, lines 17-19.

An obviousness rejection is not proper when the cited references teach away from the claimed

invention. The Federal Circuit said "[u]nder the proper legal standard, a reference will teach

away when it suggests that the developments flowing from its disclosures are unlikely to

produce the objective of the applicant's invention." Syntex (U.S.A.) LLC v. Apotex Inc., 407

F.3d. 1371 (Fed. Cir. 2005). Muroyama explicitly states that the reason for removing the metal

oxide is to provide for more reliable growth of the carbon film. Muroyama is clearly teaching

away from using an organometallic oxide as the catalyst if one wishes to obtain reliable growth

of the carbon film.

This same legal standard for teaching away was used in Takeda Chemical Industries

Ltd. V. Alphapharm Pty., 83 U.S.P.Q. 2d 1169 (Fed. Cir. 2007) where the Federal Circuit found

that the prior art article was found to teach away from the patented compound because the

negative properties would have directed one of ordinary skill away from selecting the

compound for further investigation, and in Inpro II Licensing, S.A.R.L. v. T-Mobil USA, Inc.,

78 U.S.P.Q. 2d 1786 (Fed. Cir. 2006) where the Federal Circuit noted that Inpro's specification

referred to prior-art serial connections as "a big drawback," therefore excluded serial

connections. Similarly, Muroyama teaches that the presence of oxides leads to unreliable

growth, thereby teaching away from selecting the organometallic of Muroyama as a catalyst

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and then oxidizing the catalyst as taught by Dai, and a *prima facie* case of obviousness has not been made.

## **Summary**

For the foregoing reasons, Appellant respectfully submit that the rejection of claims 1-19 is clearly erroneous. Reversal of the final rejection of claims 1-19 is respectfully requested.

Respectfully submitted,

Avetik Harutyunyan

Dated: September 29, 2008 By: /Narinder Banait/

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